Alternative Energy for Transportation: Hydrogen and Fuel Cells

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Lab-aids EXPE

EXPERIENCING SCIENCE



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Publisher: LAB-AIDS, Inc.: <u>lab-aids.com</u>

Curriculum Website: sepuplhs.org/hydrogen

Connecticut Sales Reps: Ruth Ann McDougall

HyTEC: Hydrogen Technology and Energy Curriculum

- Funded by U.S. Dept of Energy
- "Introduction to Alternative Energy: Hydrogen Fuel Cells"
- Developed by a team of scientists, engineers, curriculum developers, teachers, and other educational leaders
- Development process includes extensive classroom testing and feedback
- High School Chemistry (or Physics & Envi. Sci.)



Partners

 Lawrence Hall of Science



Schatz Energy Research Center



AC Transit



FilmSight Productions

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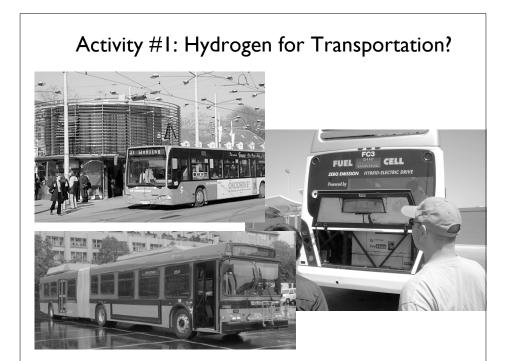
Teachers and students from SF Bay Area, Washington, Ohio, California, Connecticut, Georgia, New York, and South Carolina

Issue-Oriented Science

- Engages students in learning science and applying it to make evidence-based decisions.
- •In most cases, does not advocate a particular decision, but does advocate the use of scientific evidence and concepts in the decision-making process.
- Encourages students to look at various sides of an issue and evaluate the trade-offs involved in a complex decision.

Hydrogen and Fuel Cells in Connecticut

- Connecticut Hydrogen Fuel Cell Coalition: A collaboration of over 30 companies and agencies
- Connecticut is one of the top 5 fuel cell states, according to Fuel Cells 2000 (along with California, New York, Ohio, and South Carolina), and home to major manufacturers, high profile installations, supportive funding policies
- Connecticut has a fuel cell bus demonstration program in Hartford

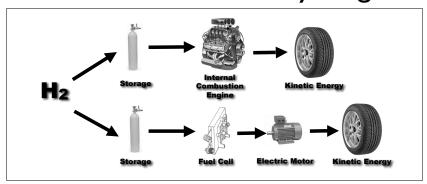


Hydrogen

- •Hydrogen is the most common element in the universe.
- •The sun is composed mostly of hydrogen gas.
- •Where is hydrogen found on Earth?
- •Hydrogen occurs naturally as a component of water, air, and hydrocarbon fuels like coal and natural gas.

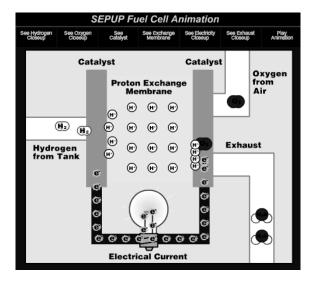
Hydrocarbon — Reformer — Hydrogen + CO₂ Water + Electricity Electrolyzer — Hydrogen + Oxygen

What do we do with Hydrogen?

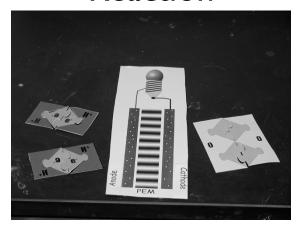


- A way to store energy (like a battery)
- A way to move energy (like electricity)
- NOT an energy source and NOT free

Activity #4: Modeling the Fuel Cell Reaction



Modeling the Fuel Cell Reaction



Now use the puzzle pieces to model what happens in the fuel cell.

Modeling the Fuel Cell Reaction







Student Activity The Fuel Cell Half Reactions

- The half-reactions:
 - Oxidation: $H_2 \rightarrow 2H^+ + 2e^-$
 - Reduction: $4H^+ + O_2 + 4e^- \rightarrow 2H_2O$
- Adding the half-reactions:
 - Oxidation: $2H_2 \rightarrow 4H^+ + 4e^-$
 - Reduction: $4H^+ + O_2 + 4e^- \rightarrow 2H_2O$

lacktriangle

 $^{\bullet}$ 2H₂+O₂ → 2H₂O + energy (electricity)

The HyTEC Curriculum *Six activities take approximately two weeks of

- instructional time.
- 1. Energy for Transportation Students examine trade-offs of various fuel/vehicle combinations.
- 2. Obtaining Hydrogen through Electrolysis - In this hands-on lab, students generate hydrogen and examine the required energy input, stoichiometry, and electrochemistry involved in the process.



The HyTEC Curriculum

3. Putting a Hydrogen Fuel Cell to Work -Students generate H_2 and O_2 , and use a single cell fuel cell to perform work.



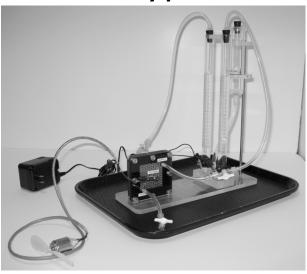
4. Modeling a Fuel Cell **Redox Reaction -**Students use model pieces and a fuel cell simulation to explore the fuel cell reaction.

The HyTEC Curriculum

- **5. Fuel Cell Efficiency** In a hands-on lab, students measure fuel cell efficiency.
- 6. Hydrogen for Transportation Students conduct research and engage in a simulated City Council Meeting to present the advantages and challenges of using hydrogen and fuel cells for a city bus program.



Prototype Kit



Website and Videos

Hydrogen Fuel Cell website: sepuplhs.org/hydrogen

- •Simulation of Fuel Cell
- •Clips from video field trip
- •Web Resources
- •Info on fuel cells





NSES Addressed

Structure of Atoms:

- •Matter is made of minute particles called atoms.
- Structure and Properties of Matter:
- •Atoms interact with one another by sharing or transferring electrons
- ·Chemical Reactions:
- •Chemical reactions occur all around us
- •Chemical reactions may release or consume energy
- •A large number of reactions involve transfer of electrons
- Catalysts lower activation energy necessary for reactions

Connecticut Standards

9-10 Core

D Inq.8 Use mathematical operations to analyze and interpret data and present relationsips between variables

D7 Describe the availability, current uses and environmental issues related to the use of hydrogen fuel cells. . .to produce electricity

D II Describe how atoms combine to form new substances by transferring or sharing electrons

Chemistry

Chemical bonds

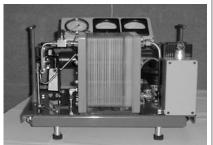
Conservation of Matter and Stoichiometry

Reaction rates (catalysts)

Applications of Fuel Cells

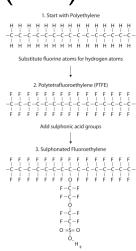
- Video highlights
- Fuel cell bus in Oakland, CA
- Portable applications: video camera, computer
- Production from hydrogen using renewable sources
- Production of hydrogen from landfill gas



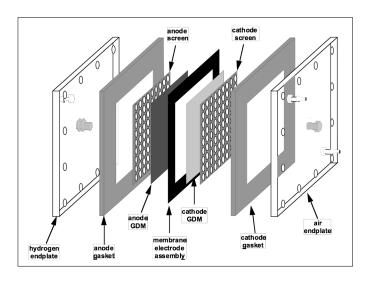


The Proton Exchange Membrane (PEM)

- Modified polyethylene hydrocarbon chains
- Fluorine substitutions create polytetrafluorethylene (PTFE: teflon®)
- To make it electrolytic: side chains with hydrophilic sulphonate (-SO₃H) groups are added



Fuel Cell Parts - Form and Function



Challenges to Hydrogen Economy



- Developing infrastructure and improving technology
- Reducing cost
- Addressing public concerns about safety
- Production of hydrogen from water using renewable energy sources

Get Involved!

- Professional Development: Berkeley, Jan. 14-15, 2011
- Contact SEPUP
 - •chris_k@berkeley.edu
 - •<u>bnagle@berkeley.edu</u>
- Power point and handouts
 - •sepuplhs.org/news.html
- Curriculum Website
 - •sepuplhs.org/hydrogen
- LAB-AIDS Booth