Teaching About Hydrogen Fuel Cells

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For More Information

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Publisher: LAB-AIDS, Inc. lab-aids.com

Curriculum Website: sepuplhs.org/hydrogen

Both #1613

HyTEC: Hydrogen Technology and Energy Curriculum

- Funded by U.S. Dept of Energy
- Module: "Investigating Alternative Energy: Hydrogen & Fuel Cells"
- High School Chemistry (or Physics & Envi. Sci.)
- Developed by a team of scientists, engineers, curriculum developers, and teachers
- Development process includes extensive classroom testing and feedback





Partners

The Lawrence Hall of Science

Schatz Energy Research Center

AC Transit

FilmSight Productions

LAB-AIDS, Inc.











Teachers and students from SF Bay Area, Washington, Ohio, California, Connecticut, Georgia, New York, and South Carolina

Issue-Oriented Science

- Engages students in learning science and applying it to make evidence-based decisions.
- Does not advocate a particular decision, but does advocate the use of scientific evidence and concepts in the decision-making process.
- Encourages students to look at various sides of an issue and evaluate the trade-offs involved in a complex decision.

Hydrogen & Fuel Cells Curriculum

Six activities take approximately two weeks of instructional time.

1. Hydrogen for Transportation? - Students examine trade-offs of various fuel/vehicle combinations.

Activity #1: Hydrogen for Transportation?



Alternative Energy Vehicles



Gasoline Internal Combustion (ICE)



Electric



Bio-diesel



Hybrid





Hydrogen Fuel Cell

Hydrogen

- Hydrogen is the most common element in the universe.
- The sun is composed mostly of hydrogen gas.
- Where is hydrogen found on Earth?

Hydrogen occurs naturally as a component of water, air, and hydrocarbon fuels like coal and natural gas.





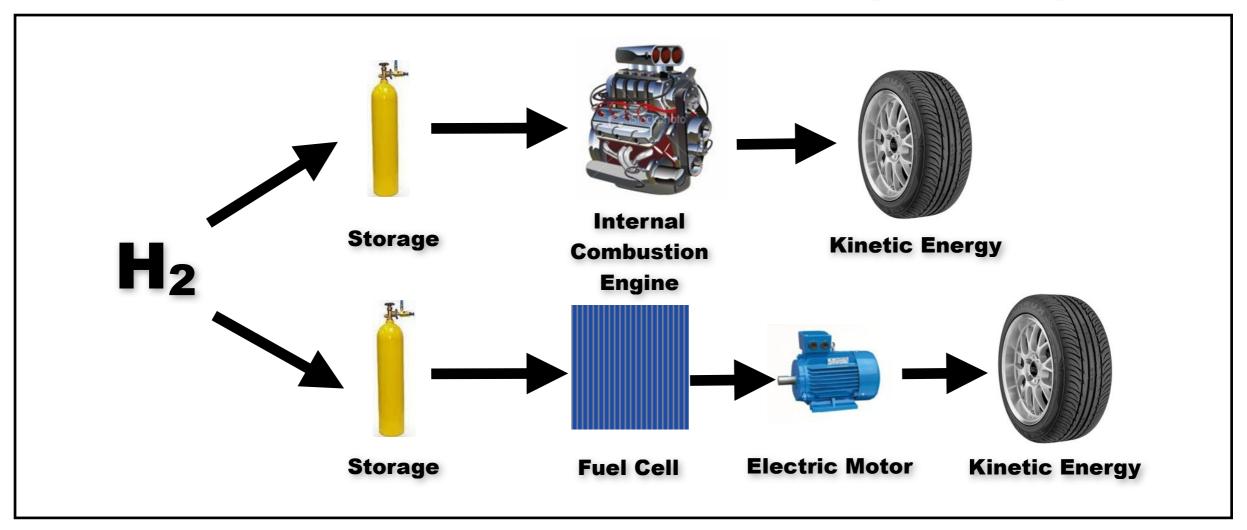






How do we get Hydrogen?

What do we do with Hydrogen?



- A way to store energy (like a battery)
- A way to move energy (like electricity)
- NOT an energy source and NOT free

Hydrogen & Fuel Cells Curriculum

- 1. Hydrogen for Transportation? Students examine trade-offs of various fuel/vehicle combinations.
- 2. Obtaining Hydrogen through Electrolysis In this hands-on lab, students generate hydrogen and examine the required energy input, stoichiometry, and electrochemistry involved in the process.

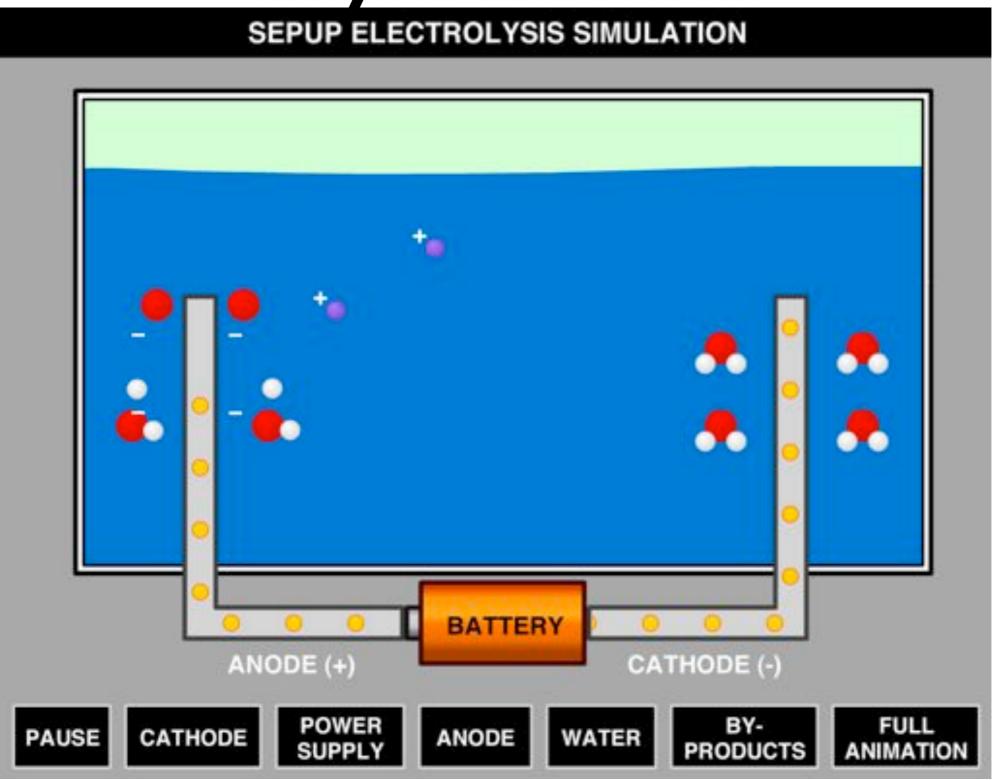


Kit Materials



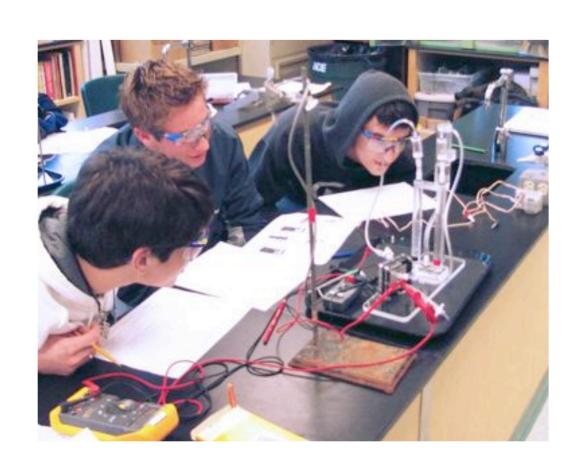
- Electrolyzer
- Power Supply
- •Fuel Cell
- Motor
- •etc.

Electrolysis Simulation



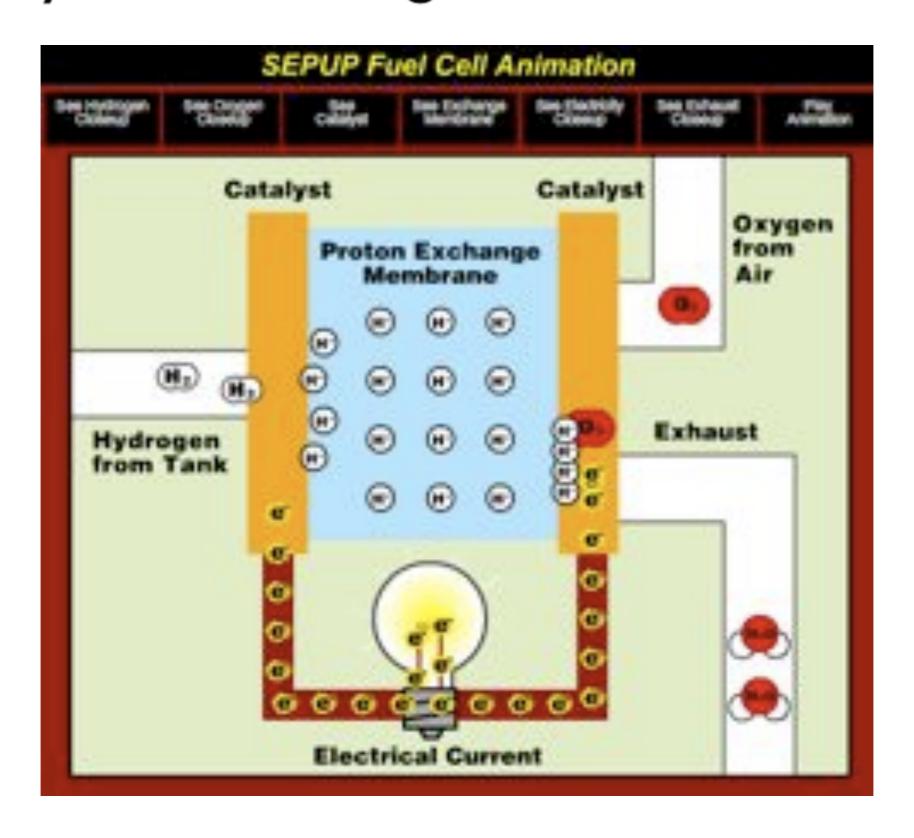
Hydrogen & Fuel Cells Curriculum

3. Observing a Fuel Cell - Students generate H₂ and O₂, and use a single cell fuel cell to perform work.



4. Modeling the Fuel Cell Reaction - Students use model pieces and a fuel cell simulation to explore the fuel cell reaction.

Activity #4: Modeling the Fuel Cell Reaction

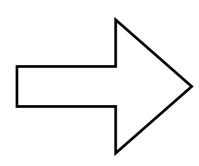


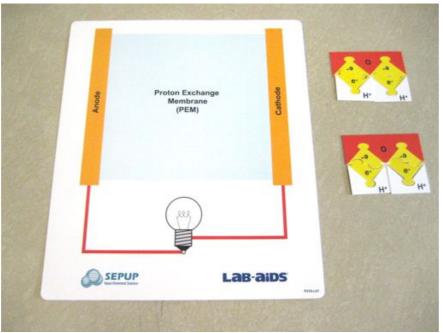
Modeling the Fuel Cell Reaction



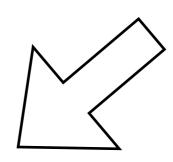
Modeling the Fuel Cell Reaction











Student Activity The Fuel Cell Half Reactions

The half-reactions:

- Oxidation: $H_2 \rightarrow 2H^+ + 2e^-$
- Reduction: $4H^+ + O_2 + 4e^- \rightarrow 2H_2O$

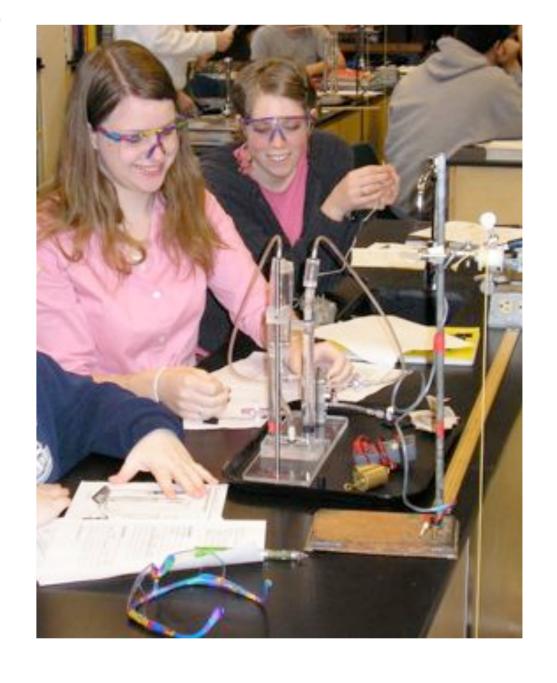
Adding the half-reactions:

- Oxidation: $2H_2 \rightarrow 4H^+ + 4e^-$
- Reduction: $4H^+ + O_2 + 4e^- \rightarrow 2H_2O$

$$2H_2 + O_2 \rightarrow 2H_2O + \text{energy (electricity)}$$

Hydrogen & Fuel Cells Curriculum

- 5. Fuel Cell Efficiency In a hands-on lab, students measure fuel cell efficiency.
- 6. Hydrogen for Buses?
 Students conduct research and engage in a simulated City Council Meeting to present the advantages and challenges of using hydrogen and fuel cells for a city bus program.



Website and Videos

Hydrogen & Fuel Cells website: sepuplhs.org/hydrogen

Simulations
Clips from video field trip
Web Resources
Info on fuel cells





Science Education for Public Understanding Program

RESOURCES

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Video 1: Intro

Video 2: On the Bus

Video 3: Portable Fuel Cells

Video 4: Solar Power

Applications of Fuel Cells

Video highlights:

- Fuel cell bus in Oakland, CA
- Portable applications: video camera, computer
- Production from hydrogen using renewable sources
- Production of hydrogen from landfill gas





NSES Addressed

Structure of Atoms:

• Matter is made of minute particles called atoms.

Structure and Properties of Matter:

Atoms interact with one another by sharing or transferring electrons

Chemical Reactions:

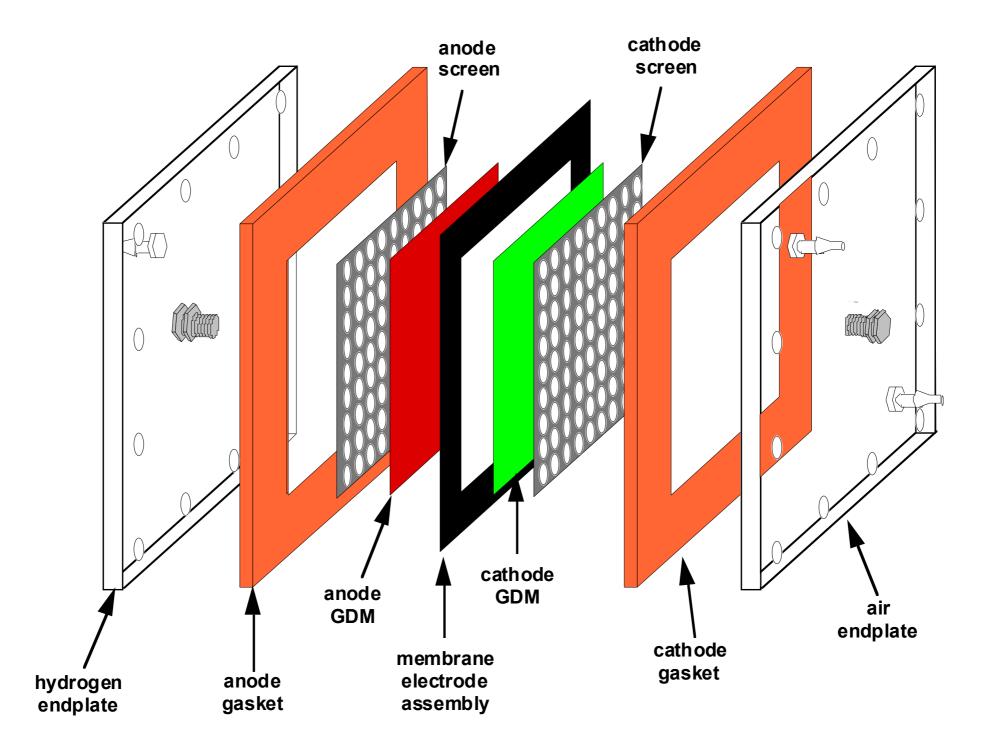
- Chemical reactions occur all around us
- Chemical reactions may release or consume energy
- A large number of reactions involve transfer of electrons
- Catalysts lower activation energy necessary for reactions

Challenges to Hydrogen Economy



- Developing infrastructure and improving technology
- Reducing cost
- Addressing public concerns about safety
- Production of hydrogen from water using renewable energy sources

Fuel Cell Parts - Form and Function



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